

Online Assessment (Even Sem/Part-I/Part-II Examinations 2019 - 2020)

Course Name - Pharmaceutical Chemistry I

Course Code - 1.2T

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Answer all the questions. Each question carry one mark.

9. 1. pH of blood is- *

Mark only one oval.

- 6.8
- 7.8
- 4.5
- 7.4

10. 2. Permissible limit of arsenic in drinking water as per I.P. is- *

Mark only one oval.

- 4 ppm
- 2 ppm
- 1 ppm
- 8 ppm

11. 3. As per I.P., refrigerated temperature means- *

Mark only one oval.

- 0°C-4°C
- 2°C-8°C
- 0°C-8°C
- None of these

12. 4. Central Pharmacopoeia laboratory is located at - *

Mark only one oval.

- Delhi
- Kolkata
- Ghaziabad
- Mumbai

13. 5. The main function of astringent is- *

Mark only one oval.

- Protein precipitation
- Reduce cell permeability
- Inhibit the trans-capillary
- All of these

14. 6. The combination of aluminum and magnesium salts as antacid are used to- *

Mark only one oval.

- To increase the activity of each antacid
- Balance the constipation and laxative effect
- Increase the absorption of each antacid
- None of these

15. 7. Milk of magnesia is a common name of- *

Mark only one oval.

- Suspension of Magnesium hydroxide
- Suspension of magnesium carbonate
- Suspension of magnesium oxide
- None of these

16. 8. Synonym of $\text{CuSO}_4, 5\text{H}_2\text{O}$ is- *

Mark only one oval.

- Blue vitriol
- Epsom salt
- Green vitriol
- Rochelle salt

17. 9. Which of the following is used as saline cathartic? *

Mark only one oval.

- NaCl
- NaHCO₃
- MgSO₄
- CaCO₃

18. 10. Which of the following indicator is used during the assay of KMnO₄? *

Mark only one oval.

- Methyl Orange
- Self-Indicator
- Mucilage of Starch
- None of these

19. 11. Which one of these inorganic salts possesses sedative action? *

Mark only one oval.

- Sodium citrate
- Potassium bromide
- Calcium phosphate
- None of these

20. 12. Which one of the following is a Lewis acid? *

Mark only one oval.

- NH₃
- NH₄⁺
- CN⁻
- None of these

21. 13. Which one of the following is used as expectorant? *

Mark only one oval.

- KI
- CuSO₄
- KBr
- FeSO₄

22. 14. Example of iodophore- *

Mark only one oval.

- Strong solution of iodine
- Lugol's solution
- Povidone iodine
- All of these

23. 15. Solubility of boric acid is increased by- *

Mark only one oval.

- Hot water
- HCl
- Glycerol
- Alcohol

24. 16. Which one of following is the natural source of Hg - *

Mark only one oval.

- Kelp
- Galena
- Cannabar
- Colemanite

25. 17. Which one of the following alum is used as an indicator in volumetric titration? *

Mark only one oval.

- Potash alum
- Ferric alum
- Ammonium alum
- All of these

26. 18. On heating zinc carbonate gives- *

Mark only one oval.

- Zinc oxide
- Zinc metal
- Zinc bicarbonate
- None of these

27. 19. Which one of these zinc salt is used in ophthalmic preparation as mild bactericide? *

Mark only one oval.

- Zinc chloride
- Zinc oxide
- Zinc stearate
- Zinc sulphate

28. 20. Which one of the following acts as an anti-emetic agent? *

Mark only one oval.

- Ammonium chloride
- Apomorphine
- Dimenhydrinate
- Diphenoxylate

29. 21. Which one of the acid is used in the treatment of skin disorders? *

Mark only one oval.

- Benzoic acid
- Propionic acid
- Acetic acid
- None of the above

30. 22. Light kaolin differs from heavy kaolin in the following aspects- *

Mark only one oval.

- Bulk density
- Purity
- Particle size
- All of these

31. 23. Chemically washing soda is- *

Mark only one oval.

- Sodium carbonate
- Sodium sulphite
- Sodium bicarbonate
- Soda lime

32. 24. Which of the following substance is used as anticoagulant? *

Mark only one oval.

- Sodium citrate
- Sodium chloride
- Sodium peroxide
- Sodium sulphate

33. 25. Sodium metal is stored in- *

Mark only one oval.

- Well closed container
- Is either kept in well closed container
- In kerosene , kept in well closed container
- In water, kept in well closed container

34. 26. Which of the following is used for the treatment of dandruff? *

Mark only one oval.

- Selenium sulfite
- Sulphur
- Yellow mercuric oxide
- None of these

35. 27. Xylenol indicator is used in - *

Mark only one oval.

- Complexometric titration
- Precipitation titration
- Redox titration
- None of these

36. 28. The lead acetate cotton wool is used in limit test for arsenic to- *

Mark only one oval.

- Make the arsenic gas
- Develop the yellow colour
- Trap the hydrogen sulphide gas
- None of these

37. 29. The combination of zinc oxide and phosphoric acid is used for- *

Mark only one oval.

- Abrasion purposes in tooth paste
- Filing dental cavities
- Prevention of dental infection
- None of these

38. 30. In the limit test for sulphate, hydrochloric acid is added to- *

Mark only one oval.

- Decrease the solubility of barium sulphate
 - Enhance the reaction of barium chloride
 - Prevent the precipitation of barium carbonate
 - None of these
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