## Online Assessment (Even Sem/Part-I/Part-II Examinations 2019 - 2020

Course Name - Pharmaceutical Organic Chemistry I Course Code - BP202T

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	Mark only one oval.
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	Bachelor of Pharmacy
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	B.SC.(CS)
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	B.SC.(ANCS)
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	B.Sc.(MM)
	B.A.(MW)
	BBA
	B.COM
	B.A.(JMC)
	BBA(HM)
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	B.OPTOMETRY
	B.SC.(MB)
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	Dip.CSE
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	MCA MCA (OC)
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	MBA
	M.SC.(BT)
	M.TECH(CSE)
	LLM
	M.A.(JMC)
	M.A.(ENG)
	M.SC.(MATH)
	M.SC.(MB)
Ar	swer all the questions. Each question carry one mark.
9.	1.Which one gives 1st position in priority table? *
	Mark only one oval.
	Mark only one oval.  Aldehyde
	Aldehyde
	Aldehyde  Ketone
	Aldehyde  Ketone  Nitro
	Aldehyde  Ketone  Nitro
10.	Aldehyde  Ketone  Nitro
10.	Aldehyde  Ketone  Nitro  Carboxylic acid
10.	Aldehyde  Ketone  Nitro  Carboxylic acid  2.Structural isomerism can be classified as types *
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10.	Aldehyde  Ketone  Nitro  Carboxylic acid  2.Structural isomerism can be classified as types *  Mark only one oval.
10.	Aldehyde  Ketone  Nitro  Carboxylic acid  2.Structural isomerism can be classified as types *  Mark only one oval.  1 3

11.	3.Which of the following is a 2° alcohol? *
	Mark only one oval.
	1-Propanol 2-Propanol Cyclohexanol 2-methyl-2-butanol
12.	4.Which of the following compound has the functional group – OH *  Mark only one oval.
	2-butanone  1, 2-ethandiol  Nitrobenzene  Ethanal
13.	5.In C=C, there is *  Mark only one oval.  sp3 hybridization  sp hybridization  sp2 hybridization  no hybridization

14.	6.Ketones are prepared by the oxidation of *
	Mark only one oval.
	Primary alcohol
	Secondary alcohol
	Tertiary alcohol
	None of these
15.	7.Markonikov's addition of HBr is not applicable to *
	Mark only one oval.
	propane
	1-butane
	1-pentene
	2-butene
16.	8.Rectified spirit is *
	Mark only one oval.
	100% ethanol
	90% ethanol
	100% methanol
	95% ethanol

17. 9.The high boiling points of alcohols,as compared to the corresponding alkanes,are due to \*

Mark only one oval.

)	Lydrogor	hondina
)	Hydroger	i bollallia

- Heavy oxygen atom
- Water solubility
- None of these
- 18. 10.As per IUPAC nomenclature, aldehyde named as \_\_\_\_\_. \*

Mark only one oval.

- alkenals
- alkanals
- alkynals
- None of these
- 19. 11.Which one is a conjugated diene?\*

Mark only one oval.

- $\bigcirc C = C C = C$
- $\bigcirc$  C C = C C
- $\bigcirc C = C C C = C$
- $\bigcirc$  C = C = C

20.	12.Select the incorrect statement regarding alkanes *	
	Mark only one oval.	
	It is otherwise known as Paraffin  It is an acyclic saturated hydrocarbon  In alkanes, C-C bonds are single  Alkanes have the general formula CnH2n	
21.	13.The oxidizing agent in the Tollen's test for the presence of an aldehyde is *	
	Mark only one oval.	
	Ag(NH3)20H Ag Cu2+ Cu20	
22.	14.Which amine is not soluble in water? *	
	Mark only one oval.	
	Methylamine Dimethylamine Trimethylamine All of these are water soluble	

23.	15.When ethyl chloride reacts with nascent hydrogen, what is the formed product?
	Mark only one oval.
	Methane
	Propane
	Butane
	Ethane
24.	16.Satzeff rule states theis formed most readily *
	Mark only one oval.
	least substituted alkane
	most substituted alkane
	least substituted alkene
	most substituted alkene
25.	17.General formula for alcohols is *
	Mark only one oval.
	CnH2n
	CnH2n+10H
	CnH
	СНЗ

20.	18. Markonikov addition "
	Mark only one oval.
	gives the most stable carbocation
	gives the least stable carbocation
	is addition to a carbon atom containing the least hydrogen atom
	none of these
27	10 The product formed in Aldel condensation is *
27.	19.The product formed in Aldol condensation is *
	Mark only one oval.
	beta-hydroxy aldehyde or a beta-hydroxy ketone
	an alpha-hydroxy aldehyde or ketone
	an alpha, beta unsaturated ester
	a beta-hydroxy acid
28.	20.Which of the following reaction can be used to prepare propane alkane? *
	Mark only one oval.
	Wurtz Reaction
	Wolf-Kishner Reaction
	Kolbe electrolysis
	All of these

29.	21.Which alkyl halide has the highest reactivity for a particular alkyl group? *
	Mark only one oval.
	R-F R-CI
	R-I
	R-Br
30.	22.Clemmensen reduction of a ketone is carried out in the presence of which of the following? *
	Mark only one oval.
	H2 and Pt as catalyst
	Glycol with KOH
	Zn-Hg with HCl
	LiAlH4
31.	23.Lindlar's catalyst is *
	Mark only one oval.
	LiAIH 4
	Pd/BaSO4 in quinoline
	NH 2NH 2
	HCI/ZnCI2

32.	24.Acid anhydrides on reaction with primary amines give
	Mark only one oval.
	amide imide secondary amine
	imine
33.	25.Which of following is a Tertiary amines? *
	Mark only one oval.
	Methylamine Dimethylamine Ethyl methylamine Trimethylamine
34.	26.Acetyl salicyclic acid is anderivative of salicyclic acid *
	Mark only one oval.
	Methyl
	acetyl
	butyl
	pentyl

35.	27.Which is the following compound show geometrical isomerism? *
	Mark only one oval.
	2-Butene
	2-Methyl-2-Butane
	1-Pentanol
	1,2-Dichloropropane
36.	28.The shape of a p orbital is *
	Mark only one oval.
	oval
	spherical
	dumb-bell
	rectangular
37.	29.Which of the following compound assigned the Octane Number of zero: *
	Mark only one oval.
	n-Octane
	2,3,3-Trimethylpentane
	n-Heptane
	2,2,4-Trimethylpentane

38.	30.Which of the following compounds reacts slowest with Lucas reagent at room temperature? *
	Mark only one oval.
	1-butanol
	2-propanol
	2-butanol
	2-methyl-2-propanol

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