Online Assessment (Even Sem/Part-I/Part-II Examinations 2019 - 2020

Course Name - Advanced Chemistry Course Code - BBT403

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Α	nswer all the questions. Each question carry one mark.
9.	1. The building block of protein is
	Mark only one oval.
	Wax
	Carbohydrate
	Amino acids
	Lipid
10.	2. Ozonolysis of an organic compound 'A' produces acetone and propionaldehyde in equimolar mixture. Identify 'A' from the following compounds
	Mark only one oval.
	1-Pentene
	2-Methyl-1-Pentene
	2-Methyl-2-Pentene
	2-Pentene

11.	3. When 2-butyne is treated with Pd-BaSO4; the product formed will be:
	Mark only one oval.
	1-butene Trans 2-butene
	Cis 2-butene 2-hydroxy butane
12.	4. In peptide each carbonyl group are held by what with H atom of –NH group
	Mark only one oval.
	Intermolecular H bonding
	Intramolecular H bonding
	Both of these
	None of these
13.	5. An example of oxygen containing amino acid is
	Mark only one oval.
	Serine
	Threonine
	Valine
	Both serine and threonine

14.	6. An example of optically inactive amino acid is
	Mark only one oval.
	Lysine Leucine Aspartic acid Glycine
15.	7COOH group of amino acid can be activated by
	Mark only one oval.
	DCC p-nitrophenol Both DCC and p-nitrophenol BOC
16.	8. An example of basic amino acid is
	Mark only one oval.
	Arginine Lysine Aspartic acid Both lysine and arginine

17.	9. The peptide bond in proteins is
	Mark only one oval.
	Planar but rotates to three preferred dihedral angles Non-polar but rotates to three preferred dihedral angles Planar and usually found in a trans conformation Not cleaved by hydrolysis
18.	10. The enzyme used to protect N terminal end of amino acid is
	Mark only one oval.
	Carboxypeptidase Lysase Glysase Leucine aminopeptidase
19.	11. In peptide , C terminal end is written on which side?
	Mark only one oval.
	Left
	Right
	Both left and right
	None of these

20.	12. Glucose on reaction with Br2 water produces
	Mark only one oval.
	Glucaric acid Gluconic acid
	Fructose
	Sucrose
21.	13. An example of non reducing sugar is
	Mark only one oval.
	Fructose
	Glucose
	Sucrose
	Mannose
22.	14. Gangloside contains branched change of multiple sugar residues. How many sugar residues are present?
	Mark only one oval.
	7
	8
	9
	10

23.	15. Ozonolysis of acetylene produces
	Mark only one oval.
	Glyoxal
	Methyl glyoxal
	Phenyl glyoxal
	Ethyl glyoxal
24.	16. Glucose is
	Mark only one oval.
	Cyclic hemiacetal
	Cyclic hemiketal
	Glycoside
	Glucosazone
25.	17. LDBVE rearrangement is the reaction between
	Mark only one oval.
	Glucose with mild alkali
	Sucrose with alkali
	Lactose with alkali
	Glucose with Concentrated Nitric acid

26.	18. Simplest carbohydrate is
	Mark only one oval.
	Dihydroxy acetone Glyceraldehyde Glucose Gulose
27.	19. Name the major storage of carbohydrate in plants Mark only one oval.
	Cellulose Chitin Starch Glucose
28.	20. Lower fatty acid contains

29.	21. Lipids are
	Mark only one oval.
	Esters of fatty acids Esters of glycerol
	Esters of fatty acid with glycerol
	Esters of butanal
30.	22. The mass of iodine absorbed in gm by 100 gm of fat is known as
	Mark only one oval.
	Saponification number
	Acid value
	RM number
	Odine number
31.	23. Identify the addition reaction which is not undergone by the alkenes
	Mark only one oval.
	Mercuration
	Hydroboration
	Halogenation
	Oxymercuration

32.	ammoniacal cuprous chloride is
	Mark only one oval.
	Ethane
	Methane
	Acetylene
	Ethylene
33.	25. Addition of HBr to propylene in presence of benzoyl peroxide, follows :
	Mark only one oval.
	Carbanion mechanism
	Bayer's rule
	Markownikoof's rule
	Anti markownikoof's rule
34.	26. Carbohydrates are known as
54.	
	Mark only one oval.
	Hydrates of Carbon
	Carbonates
	Glycolipids
	Phospolipids

35.	27. A polysaccharide formed by β1-4 glycosidic linkage is
	Mark only one oval.
	Starch Lysose Sucrose Cellulose
36.	28. Green house gas is
	Mark only one oval.
	Carbon monoxide Sulfur dioxide Hydrogen sulphide All of these
37.	29. Ozone hole in Antarctica is due to Mark only one oval.
	Methane Carbon dioxide CFC Nitrogen

38.	30. d-glucose and I-glucose are
	Mark only one oval.
	Stereoisomers
	Anomers
	Keto-aldose isomers
	Geometrical isomers

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