



Green Synthesis of Ni-Fe co-doped TiO₂ (Ni_{0.05}Fe_{0.05}Ti_{0.9}O₂) Nanoparticles for Enhanced Supercapacitor Applications

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This work reports the green synthesis of Ni and Fe co-doped TiO₂ nanocomposites *via* solution combustion using aloe vera gel as bio-fuel. The prepared Ni_{0.05}Fe_{0.05}Ti_{0.9}O₂ composition was confirmed by XRD and Raman spectroscopy, which showed anatase phase retention and successful substitution of Ti⁴⁺ by Ni²⁺ and Fe³⁺, inducing lattice distortion and local symmetry defects. SEM revealed reduced particle size and improved dispersion due to co-doping. Electrochemical studies including cyclic voltammetry, galvanostatic charge-discharge and impedance spectroscopy demonstrated enhanced capacitance, energy density and charge transfer dynamics in NiFeTiO₂ compared to pure and Fe-doped TiO₂. The NiFeTiO₂ electrode achieved a high specific capacitance (140 F g⁻¹), energy density (280 Wh/kg) and power density (54 kW/kg), highlighting the complementary role of binary doping in improving conductivity and pseudocapacitive behaviour. Overall, green synthesis and transition metal co-doping offer a sustainable route to high-performance supercapacitor electrode materials.

Keywords: Green synthesis, Aloe Vera, Energy storage, Supercapacitor.