

Effect of role of industrially important solvents on excess thermodynamic functions of non-electrolyte solutions

N Hari Krishna^{*a}, K M Prasuna^b, T Baby^c, P Venkateswarlu^d & B Ramachandra^c

^aDepartment of Chemistry, Annamacharya Institute of Technology and Sciences, Tirupathi 517 520, A. P., India

^bDepartment of BS&H, Siddhartha Institute of Engineering Technology, Puttur 517 583, A. P., India

^cDepartment of Chemistry, Annamacharya Institute of Technology and Sciences, Tirupathi 517 520, A. P., India

^dDepartment of Chemistry, S.V University, Tirupati 517 502, A. P., India

E-mail: harinara052@gmail.com, ponneri.venkateswarlu@gmail.com

Received 21 August 2024; accepted (revised) 24 June 2025

The density (ρ) with displacements (u) for mixtures of binary compounds containing 1,3-dichlorobenzene containing *o*-, *m*-, and *p*-cresol have been experimentally determined and has an expression within the temperature span of 303.15, 308.15 and 313.15K in addition to over the entire mixture span. The evaluation for all aqueous integrals has been utilized for calculating surplus volumes (V^E), surplus isotropic compressibility (K_s^E), along with isotropic compressible material (K_s). The Redlich-Kister expression has been fitted using the estimated surplus variables. The outcomes have been studied in the light of chemical mechanisms and structural implications among individual compounds. Various theoretical models have been employed for evaluating the obtained aqueous integrals based on noise data, particularly a collision aspect hypothesis. For identical configurations, compressibilities and surplus fractional molecular quantities have been evaluated.

Keywords: Binary integrating, Surplus attributes, Theory, Accident variables, 1,3-Dichlorobenzene, *o*-, *m*-, and *p*-cresol