

## **BRAINWARE UNIVERSITY**

## Term End Examination 2021 - 22 Programme – Bachelor of Pharmacy Course Name – Advanced Instrumentation Techniques Course Code - BP811ET (Semester VIII)

Time allotted: 1 Hrs.30 Min.

Full Marks: 75

[The figure in the margin indicates full marks.]

## Group-A

(Multiple Choice Type Question)

1 x 75=75

| (Mutiple   | Choice Type Question)                           |
|--|---|
| Choose the correct alternative from the J                          | following:                                      |
| (1) NMR Spectroscopy is used for determinates?                     | nining structure in which of the following mate |
| a) Radioactive materials   | b) Insoluble chemical compounds                 |
| c) Liquid  | d) Gases.                                       |
| (2) During the limit of stray light which of                       | chemical is used?                               |
| a) Calcium Chloride  | b) Sodium Chloride                              |
| c) Magnesium Chloride  | d) Potassium Chloride                           |
| (3) In Ultraviolet spectrophotoscopy Lam                           | abert law deals with?                           |
| a) pH of the solution  | b) Concentration of solution                    |
| c) Appearance of solvent   | d) Thickness of cuvette                         |
| (4) During the control of absorbance which                         | ch chemical is used?                            |
| a) Calcium dichromate  | b) Sodium dichromate                            |
| c) Magnesium dichromate  | d) Potassium dichromate                         |
| (5) Calibration of High performance liqui                          | id chromatography by which chemical?            |
| a) Methanol  | b) Acetonitrile                                 |
| c) Theophylline  | d) Caffeine                                     |
| (6) During Calibration robustness of High<br>arameter is not used? | n performance liquid chromatography which p     |
| a) System to system variation                                      | b) Flow rate variation                          |
| c) Person to person variation                                      | d) Linearity variation                          |
| (7) During electronic balance calibration                          | the 1st sten we check is?                       |

| a) Verified balance weights  | b) last calibration data   |
|--|--|
| c) last validation data  | d) Bubble balance  |
| (8) At the lowest level the sample start to do   | etection its concentration called as?  |
| a) Accuracy  | b) limit of quantification   |
| c) Precision   | d) limit of detection  |
| (9) Baseline correction is the primary step a nstrument?   | fter which we run our sample in particular i   |
| a) Nuclear Magnetic Resonance  | b) Fluorimetry   |
| c) X-ray diffraction   | d) High Performance Liquid Chromatography  |
| (10) Ultraviolet spectrophotometer Cuvette a   | re made up of?   |
| a) steel and carbon  | b) plastic and aluminum  |
| c) thermocol and fiber   | d) quartz and glass  |
| (11) Which inert gas is used for gas chromate  | ography?   |
| a) nitrogen  | b) argon   |
| c) helium  | d) all the above   |
| (12) In IR spectrophotometry the solid sampl   | e run with the help of?  |
| a) magnesium bromide   | b) sodium chloride   |
| c) silver iodide   | d) potassium bromide   |
| (13) Which is mainly required for solid phase  | e extraction?  |
| a) filter paper  | b) micron filter   |
| c) holmium filter  | d) cartridge filter  |
| 14) liquid liquid extraction occurs with the h   | nelp of?   |
| a) glass funnel  | b) buchner funnel  |
| c) manifold  | d) separating funnel   |
| 15) NMR is the study of absorption of  | by nuclei in a magnetic field?   |
| a) Radio frequency radiation   | b) Radioactive radiation   |
| c) IR radiation  | d) Microwaves.   |
| 16) NMR Spectrometer provides  |  |
| mining structure in soluble chemical cor   |  |
| a) Accurate, destructive   | b) Accurate, non-destructive.  |
| c) Inaccurate, Destructive   | d) Inaccurate, Nondestructive.   |
| 17) NMR spectroscopy indicates the chemic positions of   | al nature of the and spatial   |
| a) Nuclei,neighbouring nuclei  | b) Electrons, Protons.   |
| c) Neutrons, electrons.  | d) Nuclei, electrons.  |
| 18) In NMR spectroscopy, which of the following magnetic field?  | owing must irradiate the spinning nuclei in s  |
| a) Perpendicular and stronger field  | b) Parallel and stronger field   |
| c) Perpendicular and weaker field  | d) Parallel and weaker field.  |
| 19) When energy is absorbed by the sample, in signal developed by which of the following the sample of the following the sample of the sample. | The state of the s |
| a) Radiofrequency detector   | b) Amplifier   |
| c) GM counter  | d) Photodetector   |

| (20) The amount of energy available in radio frequency of the following?   | ency radiation is sufficient for which  |
|--|---|
| a) Excite an atom  | b) Vibrate an atom  |
| c) Affect the nuclear spin of an atom  | d) Vibrate a molecule   |
| (21) The amount of energy available in radio frequ of the following?   | ency radiation is sufficient for which  |
| a) Excite an atom  | b) Vibrate an atom  |
| c) Affect the nuclear spin of an atom  | d) Vibrate a molecule   |
| (22) If the number of proton or neutrons is even the he following?   | e spin of the nucleus will be which of t  |
| a) Integral Spin   | b) Zero spin.   |
| c) Half integral spin  | d) Positive spin.   |
| (23) Signal splitting in NMR arises from   |   |
| a) Spin-spin coupling  | b) Shielding effect   |
| c) Deshielding effect  | d) Spin-Spin decoupling   |
| (24) In proton NMR spectroscopy, hydrogen bondi  |   |
| a) Shielding effect  | b) Peak splitting   |
| c) Deshielding effect  | d) All of these   |
| (25) Which of the following environmental effect i   | The safe in the safe state of |
| a) Spin Splitting  | b) electronegativity  |
| c) Shielding   | d) Chemical shift   |
| (26) Compound A has greater shielding constant th<br>have more chemical shift?   | an compound B. Which of them will   |
| a) Compound B  | b) Compound A   |
| c) Both will have equal chemical shift   | <ul> <li>d) Chemical shift has no relation with shielding constants.</li> </ul>   |
| (27) Which of the following statement is false for a   | mass spectroscopy?  |
| <ul> <li>a) Mass spectroscopy is used to identify unkn<br/>own compounds within a sample, and to el<br/>ucidate the structure and chemical properti<br/>es of different molecules</li> </ul> | b) Particle are characterized by their mass to harge ratios (m/z) and relative abundances   |
| <ul> <li>c) This technique basically studies the effect<br/>of ionizing energy on molecules</li> </ul>   | d) This technique can be used on all state of matter  |
| (28) What are the main criteria on which mass spec   | ctrometer used for?   |
| a) Composition in sample   | b) Relative mass of atoms   |
| c) Concentration of elements in the sample   | d) Properties of sample   |
| (29) Which species of the following is used to bom<br>spectroscopy has been performed?   | bard with the sample for which mass   |
| a) Alpha particles   | b) Neutrons   |
| c) Electrons   | d) Protons  |
| (30) Separation of ions in mass spectrometer take pwing?   | place on the basis of which of the follo  |
| a) Mass  | b) Charge   |
| c) Molecular weight  | d) Mass to charge ratio   |
|  |   |

| (31) What is the use of Mass spectroscopy                                      |  |
|--|--|
| a) Determination of molecular weight   | b) Elucidating the chemical structures of mole cules   |
| c) Both a &b   | d) None of the above   |
| (32) The highest m/z peak in mass spectrum is called                           | ed as  |
| a) Parent Peak   | b) Fragment peak   |
| c) Isotopic peak   | d) Base peak   |
| (33) The procedure for mass spectroscopy starts wi                             | th which of the following processes?   |
| a) The sample is bombarded by electron beam                                    | b) The ions are separated by passing them int o electric and magnetic field                        |
| c) The sample is converted into gaseous state                                  | d) The ions are detected   |
| (34) Which of the following ions pass through the                              | slit and reach the collecting plate?   |
| a) Negative ions of all masses   | b) Positive ions of all masses   |
| c) Negative ions of specific mass  | d) Positive ions of specific mass  |
| (35) Thermal analysis is defined as  | * 112  |
| a) Measurement of concentration of materials as a function of temperature      | b) Measurement of solubility of materials as a function of temperature                             |
| c) Measurement of physical properties as a function of temperature             | <ul> <li>d) Measurement of line positions of crystals a<br/>s a function of temperature</li> </ul> |
| (36) What are the two main techniques for thermal                              | l analysis?  |
| a) FTG AND DGG   | b) MSP AND FCT   |
| c) TGA AND DTA   | d) TSA AND DGF   |
| (37) Which of the following statements given belo                              | ow is false?   |
| a) TGA, DTA and DSC are measured using s<br>ame instrument                     | <ul> <li>b) TGA and DTA can be carried out simultan<br/>eously.</li> </ul>                         |
| c) TGA, DTA and DSC are measured using di fferent instruments.                 | d) TMA is a recent name of Dilatometry   |
| (38) The Ti and Tf temperature depends on which                                | of the following factor?   |
| a) Cooling rate  | b) Mechanical property of the material   |
| c) Thermal expansion coefficient   | d) Atmosphere above the sample   |
| (39) The Ti and Tf temperature depends on which                                | of the following factor?   |
| a) Cooling rate  | b) Mechanical property of the material   |
| c) Thermal expansion coefficient   | d) Atmosphere above the sample   |
| (40) What is the temperature required for the deco                             | omposition of CaCO3 in degree Celsiu   |
| s?   |  |
| a) 200   | b) 500   |
| c) 900   | d) 1200  |
| (41) On studying the reversible process during D' on both heating and cooling? |  |
| a) Esterification  | b) Hysteresis  |
| c) Methylation   | d) Carboxylation   |
| (42) In the application of DTA and DSC which o or the glasses?                 | f the following parameters is measured f   |

| a) Concentration                     |  | b) Solubility of the glass                                     |
|--------------------------------------|--|--|
| c) Cooling temper                    | rature   | d) Transition temperature                                      |
| (43) For the decomposition ps occur? | osition of the anhydrous calc  | ium oxalate, which of the following ste                        |
|                                      | ransition state, product   | b) Intermediates, anhydrous oxalate, calcium oxysalts          |
| ydroxides                            | iqueous hydrates, calcium h  | d) Intermediates, anhydrous calcium oxalate, calcium carbonate |
| (44) Which of the fol                | lowing parameters can be us-   | ed, using the DSC and DTA cells?                               |
| a) Catalytic proper                  | ties of enzyme   | b) Elasticity of crystals                                      |
| c) Enthalpy of sub                   |  | d) Line positions of phases                                    |
| (45) Collimators used                | in XRD are made up of  | present of phases  |
| a) Thin quartz tube                  |  | b) Thin Glass plates   |
| c) Thin metal plate                  | S  | d) All of these  |
| (46) As the applied vo               | oltage increases, the minimum  | m wavelength of X-radiation from a m                           |
| a) Increases                         |  | b) Decreases   |
| c) Remains same                      |  | d) Variable with metal   |
| (47) As the atomic nur               | mber increases, the frequency  | y of X-radiation from a target material                        |
| a) Increases                         | S. D. Salaran Carreston  | b) Decreases   |
| c) Remains same                      |  | d) Variable with metal   |
| (48) The detector in X               | -ray diffraction that detects the  | he visible radiation is  |
| a) Proportional Cou                  | inter  | b) Silicon diode   |
| c) Golay detector                    |  | d) Scintillation counter                                       |
| (49) Which of the follo              | owing detector in X-ray diffra   | action has more resolution                                     |
| a) Proportional Cou                  | nter   | b) Silicon diode   |
| c) Goniometer                        |  | d) Scintillation counter                                       |
| (50) Which of the follo              | wing techniques can be cons  | idered as finger spectro                                       |
| a) NMR                               | The Control of the Co | b) X-ray diffraction   |
| c) Mass                              |  | d) Fluorimetry   |
| (51) The useful range o              | f x-ray region for Pharmaceu   | utical analysis is   |
| a) 0.01 to 0.05nm                    |  | b) 0.05-0.07nm   |
| c) 0.01-0.1nm                        | -  | d) 0.2-0.3nm   |
| (52) In x-ray Powder ca              | mera technique, the detector   | nsed is  |
| a) Photographic film                 | in a more acceptance   | b) PMT   |
| c) Bolometer                         |  | d) Thermistor  |
| (53) One of the widely t             | used target material for gener   | ration of X-ray spectral line is                               |
| a) Zinc                              | general for gener  |  |
| c) Xenon                             |  | b) Molybdenum  |
| (54) X-ray diffractomete             | ers provide  | d) Manganese   |
| sent in a solid samp                 | ile.   | formation about the compounds pre                              |
| a) Quantitative                      |  | b) Qualitative   |
| c) Quantitative and qu               | 111111111111111111111111111111111111111  | d) Either quantitative or qualitative                          |
|                                      |  |  |