

## **BRAINWARE UNIVERSITY**

## Term End Examination 2021 - 22 Programme – Bachelor of Pharmacy Course Name – Pharmaceutical Organic Chemistry III Course Code - BP401T (Semester IV)

Time allotted: 1 Hrs.30 Min. Full Marks: 75 [The figure in the margin indicates full marks.] Group-A (Multiple Choice Type Question) 1 x 75=75 Choose the correct alternative from the following: (1) Which among the following correctly defines Diastereomer? a) These have same magnitude but different sign b) Nonsuperimposable object mirror relationship s of optical rotation d) Separation is very difficult c) These differ in all physical properties (2) Which of the following compounds will exhibit cis-trans isomerism? a) 2-butene b) 2-butyne c) 2-butanol d) butanal (3) An isomer of ethanol is: a) methanol b) diethyl ether c) acetone d) dimethyl ether (4) How many optically active stereoisomers are possible for butane-2,3-diol? a) 1 b) 2 c) 3 (5) Which of the following hydrocarbons does not have isomers? b)  $C_6H_{14}$ a)  $C_7H_{16}$ c)  $C_5H_{10}$ d)  $C_3H_8$ (6) How many aromatic isomers of dibromobenzene exist? a) 2 b) 3 c) 4 d) 6 (7) Which of the following does NOT exhibit geometric isomerism?

(8) Which of the following compounds displays optical isomerism?

b) 2-pentene d) 2-hexene

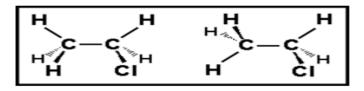
a) 1-hexene

c) 3-hexene

a) $CH_2(OH)$ - $CH_2(OH)$	b) CH <sub>3</sub> -CHCl-COOH
c) CH <sub>2</sub> =CHCl	d) CHCl=CHCl
(9) Which of the following compounds will be	optically active
a) Propanoic acid	b) 3- chloropropanoic acid
c) 2- chloropropanoic acid	d) 3-chloropropene
(10) The isomers of the substance must have	, <u> </u>
a) same chemical properties	b) same molecular weight
c) same structural formula	d) same functional group
(11) Compounds with the same molecular formu	, <u> </u>
a) Alkoxides	b) Iso compounds
c) Isomers	d) Ortho compounds
(12) If position of functional group varies in each	-
a) position isomerism	b) functional group isomerism
c) chain isomerism	d) all of them
(13) If different functional groups are present it is	
a) position isomerism	b) functional group isomerism
c) chain isomerism	d) all of them
(14) Types of structural isomerism are	
a) position isomerism	b) functional group isomerism
c) chain isomerism	d) all of them
(15) In ethane and cyclohexane which one of the ble	following pairs of conformations are more sta
a) Eclipsed and chair conformations	b) Staggered and chair conformations
c) Staggered and boat conformations	d) Eclipsed and boat conformations
(16) Lactic acid shows which type of isomerism'	?
a) Geometrical isomerism	b) Tautomerism
c) Optical isomerism	d) Metamerism
(17) Which of the following is a not a five members	pered ring?
a) Pyridine	b) Pyrrole
c) Furan	d) Thiophene
(18) Five membered rings come under which cat s of chemical behavior?	egory of heterocycle classification on the basi
a) excessive heterocycle	b) deficient heterocycle
c) equivalent heterocycle	d) Can't say about the five membered rings
(19) What is the greatest angle in pyridine ring?	
a) 116°	b) 140°
c) 124°	d) 118°
(20) Which element is present as hetero atom in	pyridine?
a) Sulphur	b) Nitrogen
c) Oxygen	d) Sulphur and nitrogen
(21) When pyrrole is treated with acetic anhydric	de then the product formed is -
a) 2-Acetyl pyrrole	b) 3-Acetyl pyrrole
c) 4-Acetyl pyrrole	d) 5-Acetyl pyrrole
(22) Furan reacts with ammonia in presence of a	lumina at $400^{0}$ C to give
1	$\sim$

a) Furfural	b) Furoic acid	
c) Pyrrole	d) Pyridine	
(23) Nitration of pyrrole is best carried out using:		
a) acetyl nitrate	b) concentrated nitric acid and sulphuric acid	
c) nitric acid	d) ammonium nitrate	
(24) Electrophilic substitution in furan usually occurs at:		
a) the o atom	b) the C(2) atom	
c) both C(2) and C(3) atom	d) the C(3) atom	
(25) Imidazole reacts with Hydrogen peroxide to give_		
a) Oxamide	b) Oxazole	
c) Oxime	d) Oxalic Acid	
(26) Imidazole is used as a		
a) Antihypertensive	b) Diuretic	
c) Antacid	d) Antipyretic	
(27) Thiazole reacts with Grignard's Reagent to produce		
a) 2-alkyl thiazole	b) 4-alkyl thiazole	
c) 5-alkyl thiazole	d) 3-alkyl thiazole	
(28) Thiazole moiety is a crucial part of		
a) Vitamin A	b) Vitamin B2	
c) Vitamin B1	d) Vitamin B12	
(29) Boiling point of Oxazole is		
a) 89 <sup>0</sup> C	b) 69 <sup>0</sup> C	
c) 29 <sup>0</sup> C	d) 109 <sup>0</sup> C	
(30) Identify the chiral molecule among the following:		
a) Isopropyl alcohol	b) 2-pentanol	
c) 1-bromo 3-butene	d) Isobutyl alcohol	
(21) What is the relationship between the struc	tung shown?	

## (31) What is the relationship between the structures shown?



- a) structural isomers
  b) geometric isomers
  c) conformational structures
  d) optical isomers

  (32) Two isomeric forms of a saturated hydrocarbon
  a) have the same structure
  b) have different compositions of elements
  c) have the same molecular formula
  d) have a different content of the isotopes of hyd rogen
- (33) Which one of the following compounds is an isomer of CH3CH2CH2CH2OH?
  - a) CH3CH2CH2OH

b) CH3CH(OH)CH3

c) CH3CH2CH2CH3

d) none of them

(34) Meso-Tartaric acid is-

a) sometimes optically active	b) always optically active
c) sometimes optically inactive	d) always optically inactive
(35) Plane polarized light is affected by	
a) Identical molecules	b) All polemers
c) Chiral molecule	d) All biomolecules
(36) Alkenes show geometrical isomerism due	to
a) Asymmetry	b) Rotation around a single bond
c) Resonance	d) restricted Rotation around a double bond
(37) Which of the following compounds may e	exist as cis trans isomers
a) 1-Butene	b) 2-Butene
c) Cyclopropane	d) Acetone
(38) Geometric isomerism is shown by	
a) Lactic acid	b) Maleic acid
c) 1-Butene	d) 1,1-Dichloroethylene
(39) The isomer of diethyl ether is	
a) (CH3)2CHOH	b) (CH3)3C-OH
c) C3H7OH	d) (C2H5)2CHOH
(40) Which of the following compounds may r	not exist as enantiomers
a) CH3CH(OH)CO2H	b) CH3CH2CH(CH3)CH2OH
c) C6H5CH2CH3	d) C6H5CHClCH3
(41) Which one of the following objects is ach	
a) Letter P	b) Letter F
c) Ball	d) A pair of hand
(42) Which of the following five membered rin	ngs is most resonance stabilized?
a) Furan	b) Thiophene
c) Pyrrole	d) Pyridine
(43) What is the dipole moment of the pyridine	
a) Zero	b) 2.2 D
c) 1.17 D	d) 4.3 D
(44) How many number of resonating structure	e stabilises a pyridine molecule?
a) 4	b) 5
c) 6	d) 7
(45) The N-atom in Pyrrole is	,
a) Sp3 hybridized	b) Sp2 hybridized
c) Sp hybridized	d) None of these
, <u> </u>	ost reactive first) of pyrrole, furan and thiophene
a) thiophene>pyrrole> furan	b) furan >pyrrole>thiophene
c) pyrrole> furan >thiophene	d) furan >thiophene>pyrrole
(47) Pyrrole is an extremly	
a) Strong Acid	b) Weak Base
c) Weak Acid	d) Strong Base
(48) Imidazole reacts with Bromine to give	
a) 4-tribromo imidazole	b) 2.4.5-tribromo imidazole

c) 2-tribromo imidazole	d) None of them
(49) Imidazole is iodinated only under alkaline condition	on to produce
a) 2-triiodoimidazole	b) 5-triiodoimidazole
c) 2,4,5-triiodoimidazole	d) None of them
(50) Boiling point of Thiazole is	
a) 117 degree C	b) 217 degree C
c) 17 degree C	d) 77 degree C
(51) Thiazole react with Sodamide to produce	
a) 2-amino thiazole	b) 4-amino thiazole
c) 5-amino thiazole	d) None of them
(52) Oxazole is	
a) Water immiscible liquid	b) Water miscible liquid
c) Partially soluble in water	d) None of them
(53) Nitration of Oxazole leads to	
a) 2-Nitrooxazole	b) 4-Nitrooxazole
c) 3-Nitrooxazole	d) 5-Nitrooxazole
(54) Pyridine react with ammonia and ethanol to produ	ice
a) 1,3-diaminopyridine	b) 2,4-dihydropyridine
c) 1,4-dihydropyridine	d) 1,4-dimethylpyridine
(55) Pyridine react with LiAlH4 to produce	
a) 1,2-dihydropyridine	b) 2,4-dihydropyridine
c) 1,4-dihydropyridine	d) None of them
(56) All carbon atom in Quinoline are	
a) sp2 hybridized	b) sp hybridized
c) sp3 hybridized	d) sp4 hybridized
(57) Quinoline react with H2SO4 at 220 degree C to pr	roduce
a) Quinoline-8-Sulphonic acid	b) Quinoline-4-Sulphonic acid
c) Quinoline-2-Sulphonic acid	d) Quinoline-6-Sulphonic acid
(58) Quinoline undergo bromination above 500 degree	C to produce
a) 2-bromoquinoline	b) 8-bromoquinoline
c) 6-bromoquinoline	d) 5-bromoquinoline
(59) Quinine is widely used as	
a) Antipyretic Drug	b) Antimalarial drug
c) Antihypertensive Drug	d) Antitubercular Drug
(60) Isoquinoline reacts with sodamide in presence of a	ammonia to produce
a) 1-aminoisoquinoline	b) 4-aminoisoquinoline
c) 6-aminoisoquinoline	d) 8-aminoisoquinoline
(61) Isoquinoline is widely used as	
a) Anaesthetic	b) Anti-inflammatory
c) Antipyretic	d) Fungicides
(62) Oxidation of isoquinoline with alkaline permanga	nate solution yields
a) Benzoic acid	b) Phthalic acid & cinchomeronic acid
c) Homophthalic acid	d) Oxalic acid
(63) All carbon and nitrogen atom in isoquinoline are	

a) sp2 hybridized	b) sp hybridized
c) sp3 hybridized	d) sp4 hybridized
(64) Acridine react with sodamide to produce	
a) 2-aminoacridine	b) 4-aminoacridine
c) 6-aminoacridine	d) 9-aminoacridine
(65) Indole soluble in	
a) Cold Water	b) Hot water
c) Insoluble in water	d) None of them
(66) Indole ring is formed by	
a) Benzene and pyyrole	b) Benzene and Pyridine
c) Pyrrole&Acridine	d) Pyrrole& Pyridine
(67) Indole react with sulfuryl chloride to produc	ce
a) 2- chloroindole	b) 3- chloroindole
c) 4- chloroindole	d) 5- chloroindole
(68) The number of tautomers of Pyrazole is	
a) Three	b) Two
c) Four	d) Six
(69) Oxazole used as a	
a) HIV inhibitor	b) Anti inflammatory
c) Muscle relaxant	d) All of them
(70) used as a reducing agent in clemme	nsen reduction
a) H2SO4	b) Ammonia
c) Hydrazine	d) Amalgameted Zinc & HCl
(71) Clemmensen reduction is always done for b	ase sensitive
a) Ketone Compound	b) Methylene Conpound
c) Carbonyl Compound	d) Acetylene Compound
(72) Cyclohexanone react with amalgameted zin	c with HCL to produce
a) Cycloalkane	b) Benzene
c) O-Cresol	d) Cycloalkene
(73) Salicylaldehyde react with zinc and HCl to	produce
a) Adipic Acid	b) Benzaldehyde
c) O-Cresol	d) Cycloalkane
(74) In Oppenauer oxidation alcohol are	oxidised much faster.
a) Secondary	b) Primary
c) Tertiary	d) Quaternary
(75) Oppenauer oxidation is used to prepare	in the pharmaceutical industry
a) Antihypertensive drug	b) Antipyretic drug
c) Analgesics	d) Anasthetic drug.